
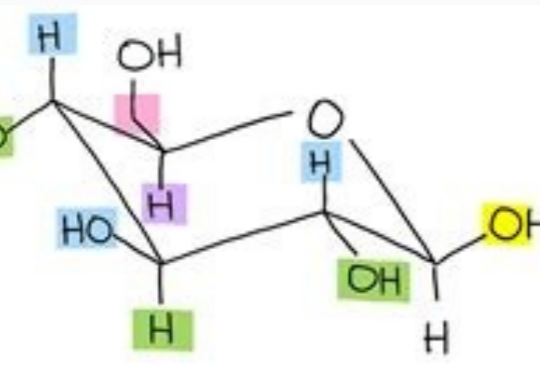
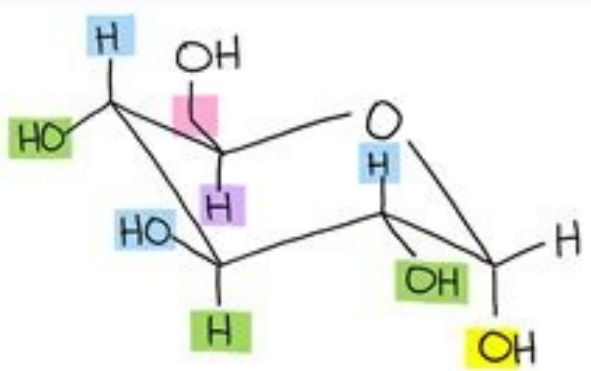
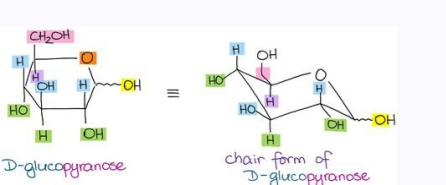
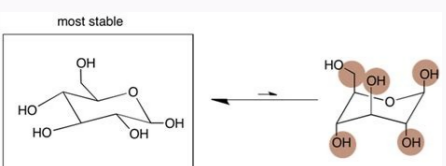
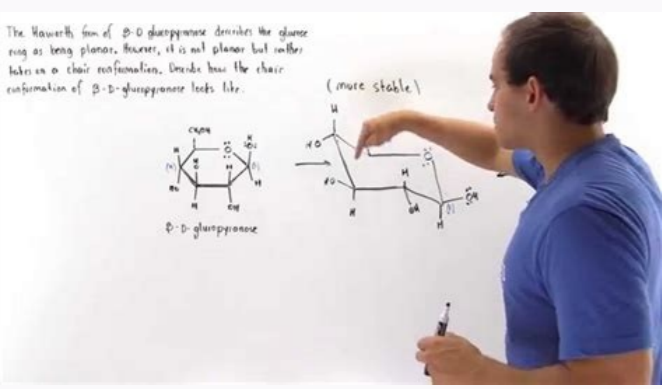


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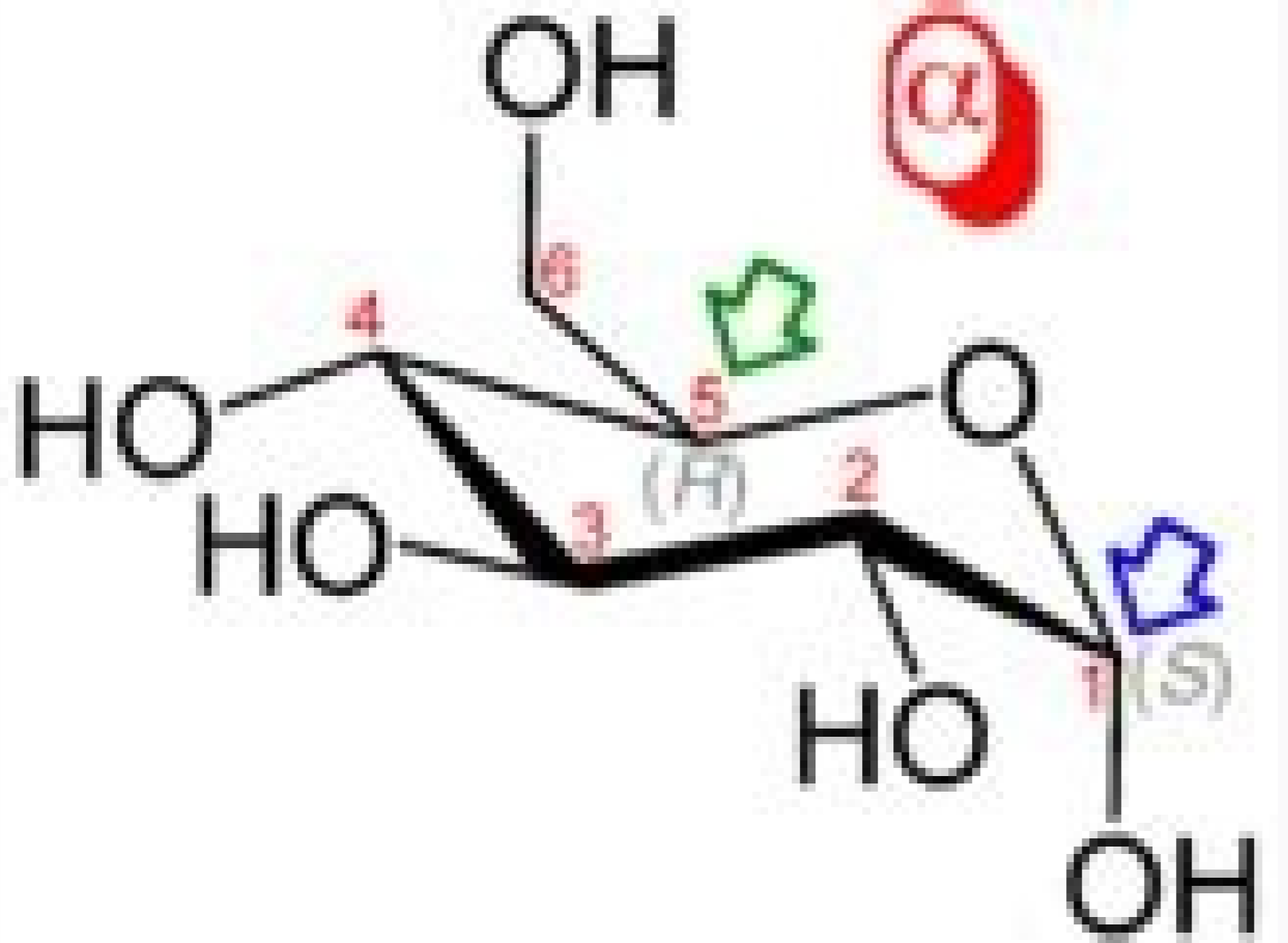
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# Alpha d glucopyranose chair conformation



$\alpha$ -D-glucopyranose

$\beta$ -D-glucopyranose



In this post I want to review the three most typical forms of carbohydrates: Fischer projection for the water open chain chains, Haworth projections focusing on cyclical pyramids, and their conformations in the regular chair. While other structures are possible (and you may have seen them before), I will focus on the chair conformation of  $\beta$ -D-glucopyranose. In particular, it will be focusing on: Aldohexoses: carbohydrates that have a hydroxyl functional group and contain six carbons. Examples of hexoses are glucose and galactose which are essential for living organisms on Earth. Pyranoses: clicked structures of six limbs resulting from haemiacetal formation between the hydroxyl group on the first carbon and the hydroxyl group on the fifth carbon. If you're not sure what exactly those terms mean, I'll go over the fundamental nomenclature of carbohydrates on this post blog here. You might want to check this out before continuing with this post so you don't get lost in the discussion when I use the terminology. Why focus specifically on the aldohexoses and pyranoses? Glucose, galactose and mannose are among the most common carbohydrates in dog living squids. Hello. In addition, the clicked shapes of six members (pyranoses) also are very important. Finally, this is a very typical kind of examination question! So you want to make sure that you really know how to convert Fischer to Haworth and chair and back. You have to know this for the MCAT even if your organic chemistry teacher does not cover it in class. You too. I will cover this peak in any introductory biochemistry class. Carbohydrates are a peak of biochemistry in the first half of the year. Fischer projection of glucose with stereo shown, as you probably know, Fischer projections are a very convenient way of drawing a molecule. Long chain chain chain with lots of chiral carbons, like a carbohydrate. In one from Fischer, we put the most oxidized red (a hydroxyl group) at the top and the rest of the chain to the right. For example, look at the structure of glucose drawn in Fischer's design form. Fischer's projection of glucose without stereochemistry shown on the left, you have a glucose structure in Fischer's projection with the stereochemistry shown. Each vertical line in Fischer's projection is far away from the observer, each horizontal line seems to be oriented towards the observer. As this is a common stereochemical convention, we don't need to show every moment for simulation. This means that Fischer's projections cannot be reversed in the space around its vertical axis, as this would make a molecule an enantiomer! So while we can turn it off  $\rightarrow$   $\leftarrow$ , it doesn't actually look like a 3D object. When it comes to the 3D representation of Fischer's projections, I really like the image of a caterpillar on its hind legs, bending back as we try to hug you. Let me illustrate what I'm trying to describe here: glucose projections, side view, I'm assuming you're familiar with Fischer's projections and everything is just an update. So let's see how you can convert Fischer's projection into Haworth's projection. Haworth projections as I mentioned in the beginning of this post I'll just focus on the pyranoses here. The pyranose is a ring of 6 members that is traditionally drawn with an anomeric carbon to the right. Haworth's projection of a pyranose in this type of representation, we assume that the lower part of the structure is oriented towards the observer. This way, we don't have to show with the wedges every moment. Stereotypical of a projection of Haworth's design projection in a different way may be ammonium. So if you have to turn in some kind of a way to maybe make a disaccharide or something, please to show the stereo! Although this may be a small thing, it may cost you a few stitches on the exam. I can tell you from the perspective of an instructor: whenever I have to guess, I get stitches in my classes. Therefore, make your life easier (and that of your instructor) and show the wedges if you draw your water ring in any form that is not a canonical/traditional representation. Steps for Converting Fischer to Haworth, to make a cyclic hemiacetal from your linear Fischer projection, you will need to follow a few simple steps: Draw a Haworth projection, number of atoms starting from the anomeric carbon and wise clockwise D, then the carbon 5 is going to be looking at sugar (down for the L) The rightness is going to be 2, 3, and 4 in the Fischer projection onto the horizontal 2, 3, and 4 in the Fischer projection onto the vertical, left-on carbons positions in the Haworth projection. 2 and 4 in the Fischer projection into Haworth projection. This sound might like a long sequence of steps, but it's actually a fairly simple process once you get a hang of it. Let's see some more examples to make it a little easier on the practice of the steps. For this exercise, I want to compare the structures of D-glucose, L-idose, and L-glucose. Using the steps above we get the following Haworth projections: D-glucopyranoside, L-idopyranoside, and L-glucopyranoside kept the same color scheme to make it a little easier to see what's going on and follow the groups as we move from one structure to another. The carbon anomaly is rich (the form is or the form is) Atomic. Now, I've defined all the stereocenters of the water open chain, except carbon anomalous rich. The anomalous carbon rich has a special place in the carbohydrate equation because it does not have a stereocenter. That means it could be facing up or down, about how the cyclization happened. While, there are factors that can do a more favorable orientation than the other, I will not go through those here to keep this post so much to the point possible. If you are curious about the stability of the anionic -OH in the "up" or "down" position, "do a little research!" (Efeito anomérico)! Positions up and down in the Haworth projection and a chair conformation looking at the glucopyranose from above, you can now easily draw the chair conformation in a general form: Haworth projection and glucopyranose chair conformation and, so easily, you can show Chair configurations specifically for  $\alpha$ -D-Glucopyranose and  $\beta$ -D-glucopyranose: alpha chair conformations and beta-D-glucopyranose, so far you can properly draw the positions of the substituents in your chair conformation, should be able to easily convert Haworth to Surprisingly, but many instructors jump the projections of Haworth. If your instructor did not cover them, I strongly recommend that you learn them, as they are useful as an intermediate step in Fischer for Haworth for chair conversions. Also you need to know those for your MCAT, DAT, PCAT, or other standard secondary papers tests. This summarizes a clean and simple Fischer conversion conversion for the Haworth projection, for the conformation of the chair for your typical aldohexoses. Occasionally, the instructors will ask you to do the opposite and convert the chair to Haworth and Fischer. So, to be ready for the test, make sure that it is practicing the steps for this way to not be picked up on examination at \*

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